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**B TECH**  
**(SEM-IV) THEORY EXAMINATION 2018-19**  
**MASS TRANSFER-I**

[Time: 3 Hours]

[Total Marks: 100]

Note: Attempt all questions.

**SECTION-A**1. Attempt **all** parts of the following:

[2\*7=14]

- a. What is Fick's law of diffusion?
- b. Define absorption factor and its significance.
- c. Discuss the types of dryers.
- d. Define Weeping, Loading in packed column.
- e. Discuss about WBT & DBT.
- f. What do you mean by Mier's supersaturation theory?
- g. Differentiate b/w molecular diffusion and knudsen diffusion.

**SECTION-B**2. Attempt any **three** parts of the following:

[7\*3=21]

- a. What is the mass transfer coefficient? Derive the relationship between overall mass-transfer coefficient and individual phase mass-transfer coefficient.
- b. A coal gas containing 2% by volume of benzene vapour enters a tray absorber at 0.250 m<sup>3</sup>/S at 26<sup>o</sup>C and 1.07\*10<sup>5</sup> N/m<sup>2</sup> total pressures for 95% removal of the benzene vapour. The solvent containing 0.005 mole fraction of benzene enters at 26<sup>o</sup>C and has an average molecular weight of 260. Calculate the circulation rate of solvent if the absorber is to be operated at 1.5 times the minimum rate. Equilibrium data-  $\frac{Y^*}{1+Y^*} = 0.125 \frac{X}{1+X}$ , where X, Y are mole ratio of benzene in solvent and absorber.
- c. Calculate the rate of diffusion of methanol at 20<sup>o</sup>C under unidirectional steady state condition through a 20 mm thick film of water when the concentrations of methanol at the opposite sides of the film are respectively 10% to 4 % methanol by weight. The diffusivity of methanol in water solution is 5.9\*10<sup>-6</sup> cm<sup>2</sup>/s. The density of 10% & 4% methanol solution at 20<sup>o</sup>C may be taken as 0.951 and 0.986 g/cc respectively.
- d. Explain mass, heat and momentum transfer analogies.
- e. Explain selection criteria for choice of solvent for absorption.

**SECTION-C**3. Attempt any **one** part of the following:

[7x1=7]

- a. In an O<sub>2</sub>-N<sub>2</sub> gas mixture at 1 std. atm 25<sup>o</sup>C, the concentration of O<sub>2</sub> at two planes 2mm apart are 10 and 20 vol% respectively. Calculate the flux of diffusion of the O<sub>2</sub> for the case where-(a) the N<sub>2</sub> is non-diffusing (b) there is equimolar counter diffusion of the two gases. Diffusivity of O<sub>2</sub> in N<sub>2</sub> is 1.81x10<sup>-5</sup> m<sup>2</sup>/s. Draw and explain the different types of equilibrium curve which may be encountered in leaching operation.
- b. Write the short note of the following –
  - (i) Film theory of mass transfer
  - (ii) Non-adiabatic humidification

4. Attempt any **one** part of the following: [7x1=7]
- It is desired to dry sheet material from 58 % to 5 % moisture content. The sheets are 0.131 m by 0.162 m by 0.071 m. The drying rate during constant rate period is 8.9 kg/m<sup>2</sup>.hr. The critical moisture content is 24.9% and the equilibrium moisture content is 1%. The sheet is to be dried from one side only and has a bone-dry density of 210 kg/m<sup>3</sup>. Determine the time required for drying. The falling rate may be assumed linear.
  - Explain the typical rate of drying curve and explain various regions in it. Also explain any one dryer equipment with neat sketch.

5. Attempt any **one** part of the following: [7x1=7]
- What is the Crystallization? Explain construction and working Swenson-Walker Crystallizer with the help of a neat sketch.
  - A hot solution containing 5000 kg Na<sub>2</sub>CO<sub>3</sub> and water with a concentration of 25 % by wt. Na<sub>2</sub>CO<sub>3</sub> is cooled to 293 K and crystal of Na<sub>2</sub>CO<sub>3</sub>.10H<sub>2</sub>O are precipitated. At 293 K, the solubility is 21.5 kg anhydrous Na<sub>2</sub>CO<sub>3</sub> per 100 kg of total water. Calculate the yield of hydrated Na<sub>2</sub>CO<sub>3</sub> crystals obtained if 5 % of the original water in the system evaporates on cooling. Also calculate the quantity of mother liquor.

6. Attempt any **one** part of the following: [7x1=7]
- A mixture of nitrogen-acetone vapors at 800 mmHg and has percentage saturation of 80%. Calculate- i) absolute molal humidity, ii) absolute humidity (kg acetone/kg nitrogen), iii) partial pressure of acetone, iv) relative humidity, v) volume percentage of acetone. Assume vapor pressure of acetone at 25<sup>0</sup>C as 190 mmHg and the Mol. Wt of nitrogen are 28.02 & acetone is 58.06.
  - Differentiate between Packed Tower and Tray Tower.

7. Attempt any **one** part of the following: [7x1=7]
- 1400 Kg (bone dry) of granular solid is to be dried under constant drying condition from a moisture content of 0.2 Kg/Kg dry solid to a final moisture content of 0.02 Kg/Kg dry solid. The drying surface is given as 0.0615 m<sup>2</sup>/Kg. Under the same conditions, the following rates were previously known. Calculate the time required for drying.

X, (kg/kg dry solid)	0.3	0.2	0.14	0.096	0.056	0.046	0.026	0.016
Rate, N (kg/m <sup>2</sup> .h)	1.71	1.71	1.71	1.46	1.29	0.88	0.54	0.376

- What is wet bulb temperature? Derive an expression for the wet-bulb depression.