

Paper Id: 

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**B.PHARM**  
**(SEM-VII) THEORY EXAMINATION 2019-20**  
**PHARMACEUTICAL ANALYSIS-III**  
**PHARMACEUTICAL ANALYSIS AND QUALITY ASSURANCE**

**Time: 3 Hours****Total Marks: 100****Note:** Attempt all Sections. If require any missing data; then choose suitably.**SECTION A**

- 1. Attempt all questions in brief. 2 x 10 = 20**
- a. What is coupling constant?
  - b. Explain the effect of hydrogen bonding on IR absorption band.
  - c. Give the principle of mass spectrometry.
  - d. Explain bathochromic and hypsochromic shift.
  - e. Give the applications of gel electrophoresis.
  - f. Differentiate between QC and QA.
  - g. What is TQM?
  - h. Give the applications of <sup>13</sup>C NMR.
  - i. Discuss fast atom bombardment ionization technique.
  - j. Define phosphorescence.

**SECTION B**

- 2. Attempt any three of the following: 10x3=30**
- a. Explain SEM and TEM.
  - b. Discuss McLafferty arrangement. Add a note on MALDI.
  - c. Illustrate the principle and applications of <sup>1</sup>H NMR.
  - d. Explain the IR, NMR and Mass spectral interpretation of ethanol.
  - e. Discuss basic concept and types of validation.

**SECTION C**

- 3. Attempt any one part of the following: 10x1=10**
- a. Illustrate Lambert-Beer law. Add a note on applications of UV spectroscopy.
  - b. Discuss instrumentation and applications of IR spectroscopy.
- 4. Attempt any one part of the following: 10x1=10**
- a. Explain principle of mass spectrometry and discuss fragmentation rules.
  - b. Explain ISO 9000 series.
- 5. Attempt any one part of the following: 10x1=10**
- a. Explain quality audit and discuss its types.
  - b. Discuss batch manufacturing record and master formula record.
- 6. Attempt any one part of the following: 10x1=10**
- a. Give the principle, instrumentation and applications of atomic absorption spectroscopy.
  - b. Give the principle, instrumentation and applications of flame photometry.
- 7. Attempt any one part of the following: 10x1=10**
- a. Discuss instrumentation of <sup>1</sup>H NMR spectroscopy. Add a note on solvents used in <sup>1</sup>H NMR spectroscopy.
  - b. What is chemical shift? Discuss factors affecting chemical shift.